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SYNTHESIS AND ANTIMICROBIAL ACTIVITY OF 4-AMINO TRINEMS

Maria Elvira Tranquillini**, Gian Luca Araldi, Daniele Donati, Giorgio Pentassuglia*,
Anna Pezzoli and Antonella Ursini

Glaxo Wellcome S.p.A., Medicines Research Centre, via Fleming 4, 37135 Verona, Italy

Abstract

This article deals with the study carried out on the synthesis of 4-amino substituted trinems 1 starting from the key-intermediate (3S, 4R)-3-[(R)-1-(t-butyldimethylsilyloxy)ethyl]-4-[(1'R, 2'S, 3'R)-1',2'-epoxycyclohex-3'-yl]azetidin-2-one 2. In particular, epoxide opening with various amines and subsequent cyclization to the corresponding trinems were explored. Their synthesis and antimicrobial activity are reported.

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 β -Lactams are the most widely used antibiotic class in the world due to their excellent therapeutic profile; however, an increasingly large number of bacterial strains is resistant to these antibiotics. In order to overcome this problem, a new class of β -lactam compounds, the trinems (formely referred to as tribactams), containing a tricyclic nucleus has recently been discovered in our laboratories.^{1,2} Amongst them, the 4-methoxy derivative GV104326 (Figure 1), shows a broad spectrum of activity, is endowed with a good enzymatic stability³ and has been selected as a candidate for development. In addition to the 4-alkoxy derivatives, the substitution with 4-amino groups has also been explored (1a-d).

FIGURE 1

$$H_3$$
C H_3 C

Considering that the relative configuration 4α , 8β represents the best compromise in terms of activity and biological stability at least for compounds studied to date, a synthetic route which permitted the stereoselective introduction of substituents at C-4 was sought. Thus, the epoxide 2 (Figure 2), synthesized by known procedures⁴, proved to be a suitable intermediate for the synthesis of the key-derivatives 3 (Figure 2).

FIGURE 2

In particular, in order to obtain the final compound 1a, sodium azide⁵ was used for the epoxide opening reaction. The reaction can be performed (Scheme 1) in the presence of either ammonium chloride⁶ or magnesium sulfate to give the azidoalcohol 4 in comparable yield (70%). Oxidation of the alcohol with pyridinium chlorochromate⁷ (70%), followed by reduction of the azido group⁶ to its corresponding amine and immediate protection with allyl chloroformate gave compound 3a (43%).

SCHEME 1

- a: NaN₃/MgSO₄7H₂O, MeOH, reflux, 20h, 70% or NaN₃/NH₄Cl, EtOH/H₂O, reflux, 20h, 73%
- b: PCC, CH₂Cl₂, 22°C, 18h, 70%
- c: 1) H₂, Pd/C10%, AcOEt, 22°C, 3 atm., 5h 2) ClCOOCH₂CH=CH₂, TEA, CH₂Cl₂, 0°C, 10min, 43%

The introduction of a secondary amino group was achieved by ring opening of epoxide 2 with primary amines as nucleophiles (Scheme 2) using ammonium chloride as catalyst. The amino alcohols 6b,c, were obtained in moderate to good yield, depending on the bulk of the nucleophile (from 100% for 6b to 45% for 6c). Sequential protection as allylcarbamates (57-75%) and oxidation with pyridinium chlorochromate (80-90%) led to compounds 3b,c.

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SCHEME 2

a: R-NH₂, NH₄Cl, EtOH/H₂O, reflux, 24h (R=Me quant.; R=cC₅H₉ 45%)

b: ClCOOCH₂CH=CH₂, 2,2,6,6-tetramethylpiperidine, CH₂Cl₂ 0°C, 10min (R=Me 75%, R=cC₅H₉ 57%)

c: PCC, CH₂Cl₂, 22°C, 9h, Y=80-90%

The same reaction conditions were also used with secondary amines (Scheme 3), such as dimethylamine. In this case a Swern oxidation was performed to give 3d in 90% yield.

SCHEME 3

a: (CH₃)₂NH, NH₄Cl, EtOH/H₂O, reflux, 3h, 55%

b: (COCl)₂/DMSO, DIPEA, , CH₂Cl₂, -55° to 22°C, 30min, 90%

Conversion of the aminoketones 3a-c into the trinem nucleus was achieved via N-1 acylation with allyloxalyl chloride followed by treatment with triethylphosphite to give a Wittig-type cyclization⁹ (Scheme 4). It is worth noting that cyclization yields decreased with increasing steric hindrance of the substituents at

the 4 position (82% in the case of 9a, 16% for 9c). Unfortunately, under the conditions used, the 4-dimethylamino trinem was not isolated unstead many degradation compounds were recovered.

Deprotection of the hydroxyethyl side-chain using tetrabutylammonium fluoride and acetic acid in tetrahydrofuran afforded compounds 10a-c (50-60%), which were deprotected under standard conditions¹⁰ to give the final derivatives 1a-c (75-85%).

SCHEME 4

- a: 1) ClCOCOOCH₂CH=CH₂, K₂CO₃, TEA, CH₂Cl₂, 22°C, 45min 2) P(OEt)₃, xylene, reflux, for 9a reflux 2h and 82%, for 9b reflux=7h and 70%, for 9c reflux=20h and 16%
- b: TBAF 3H₂O, AcOH, THF, 22°C, 15h, 50-60%
- c: Pd(PPh₃)₄, dimedone, THF, 22°C, 1h, 75-85%

Compound 1d was prepared via methylation of the 4-methylamino substituted trinem 1b in 30% yield (Scheme 5).

SCHEME 5

a: Me₂SO₄, phosphate buffer pH 8.5, THF, 22°C, 4h, 30%

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Antibacterial activity was evaluated against a series of Gram positive and Gram negative bacterial strains. Table 1 reports the *in vitro* antibacterial activity 11 (MIC µg/ml) of compounds 1a-d in comparison with GV104326, as determined by the microtiter broth dilution test (MIC values). All the compounds were active against both Gram positive and Gram negative bacteria, with improved activity against *Pseudomonas aeruginosa* with respect to GV104326. Furthermore, these compounds showed good stability both to β -lactamases and to renal human dihydropeptidase.

Compound	S. a. 663	S.a. 853	E.faecalis 850	E. coli 1850	E. coli 1919	P.aeruginosa 1911
GV104326	0.25	0.25	1.00	0.50	0.50	>32.00
1a	0.25	0.25	16.00	1.00	0.50	4.00
1b	0.25	0.50	16.00	2.00	0.50	4.00
1c	8.00	4.00	32.00	8.00	32.00	32.00
1d	0.25	0.25	>32.00	2.00	1.00	8.00

TABLE 1

In conclusion, the stereoselective synthesis of the 4-amino trinems was successfully achieved starting from the homochiral epoxy derivative 2. The epoxide ring opening occurred in satisfactory overall yield. The next critical step was the cyclization of the key-interemediates 3 to trinems, which was dependent on the steric hindrance of the substituent at the 4 position. Cyclization of compound 3d using this route failed, nevertheless, the synthetic problems were overcome by preparing 1d by direct methylation of 1b.

All these compounds show good activity against both Gram positive and Gram negative strains; in particular, compounds 1a and 1b show increased antibacterial activity versus *Pseudomonas aeruginosa*.

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- # to whom correspondence should be addressed. E-mail address: met7464@ggr.co.uk Fax:(39) 45 9218196
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